

# Nanotechnology & Colorants for Coatings

Vinayak Natu

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- What is the crux of nanotechnology?
- How do nanopigments differ in properties?
- Which are the novel nano colorants?

What is the potential of nanotechnology ?

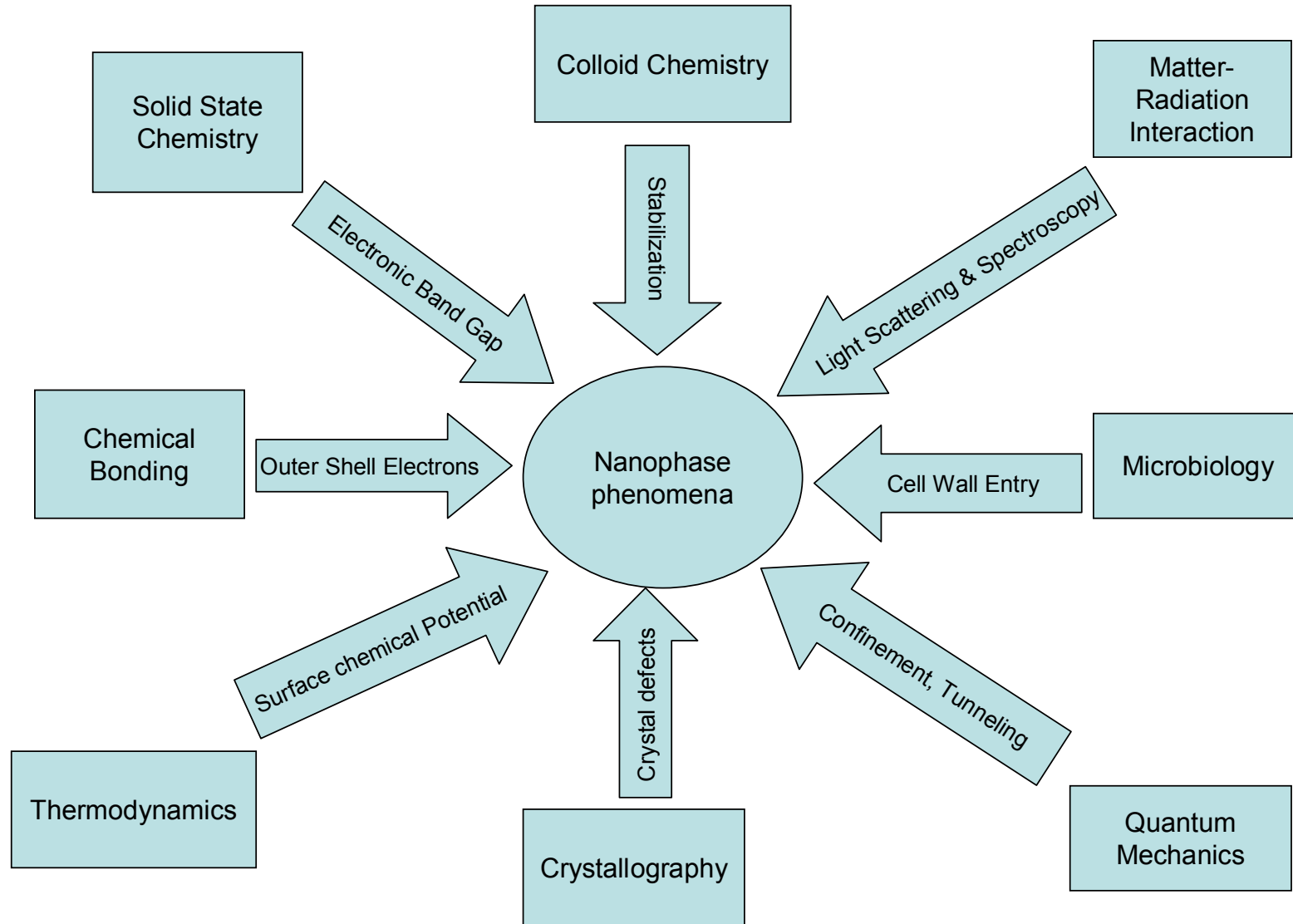
# Nanotechnology: Myths

1. It is a distinct field of technology
2. Fine particle technology = Nanotechnology
3. It is unsuitable for products of common use
4. The methods of production are too sophisticated & expensive

# Potential

- Breakthrough possibilities in chemistry to fabricate materials with novel mechanical, electrical, thermal, optical, biological properties.
- Nano-industrial revolution like semiconductors
- Will create millions of jobs in next 10 yrs
- NNI of US: 2009 Budget is US \$1.5

# Interdisciplinary Nature of Nanotechnology



What is the crux of nanotechnology?

# Criteria for Nanomaterials

1. The basic structural unit only a few atoms large size along at least one dimension:

‘Quantum Confinement’ ( 5~50 nm)

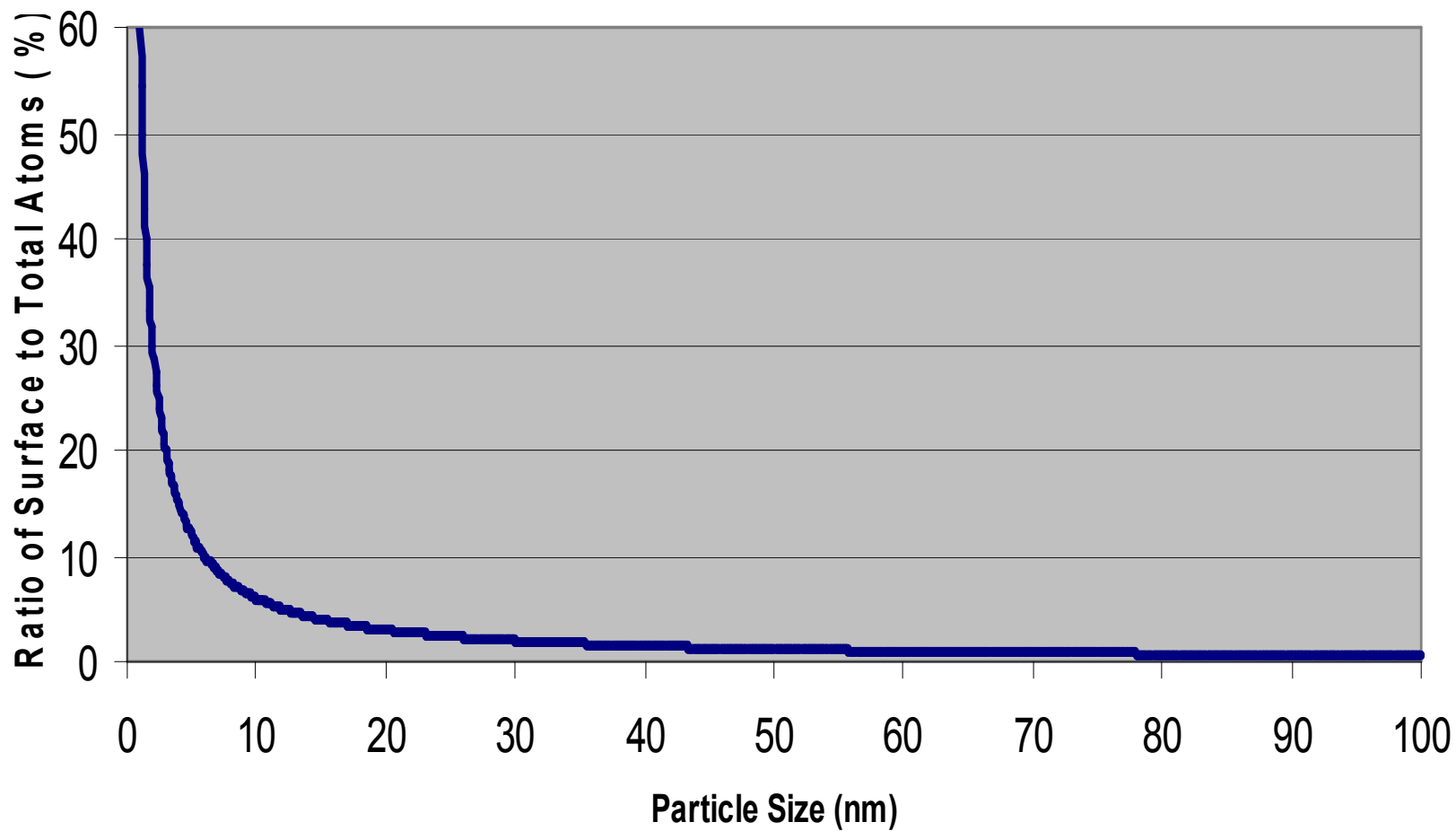
2. Properties distinctly different from the bulk material:

Mere extrapolation of properties, as the particle size decreases into the nanometer region, is not enough for the material to be termed as nanomaterial. Point of inflection is essential.

# What Changes at Nanoscale?

1. Number of atoms( molecules) on the surface becomes significant
2. Intrinsic properties become extrinsic
3. Crystalline matter becomes defect free
4. Gravity is inconsequential
5. Polymer particles become almost monomolecular
6. Particles become invisible
7. Mie scattering → Raleigh Scattering
8. Photocatalytic activity changes drastically
9. Particles can enter cell wall

**Ratio of Surface to Total Atoms vs. Particle Size**



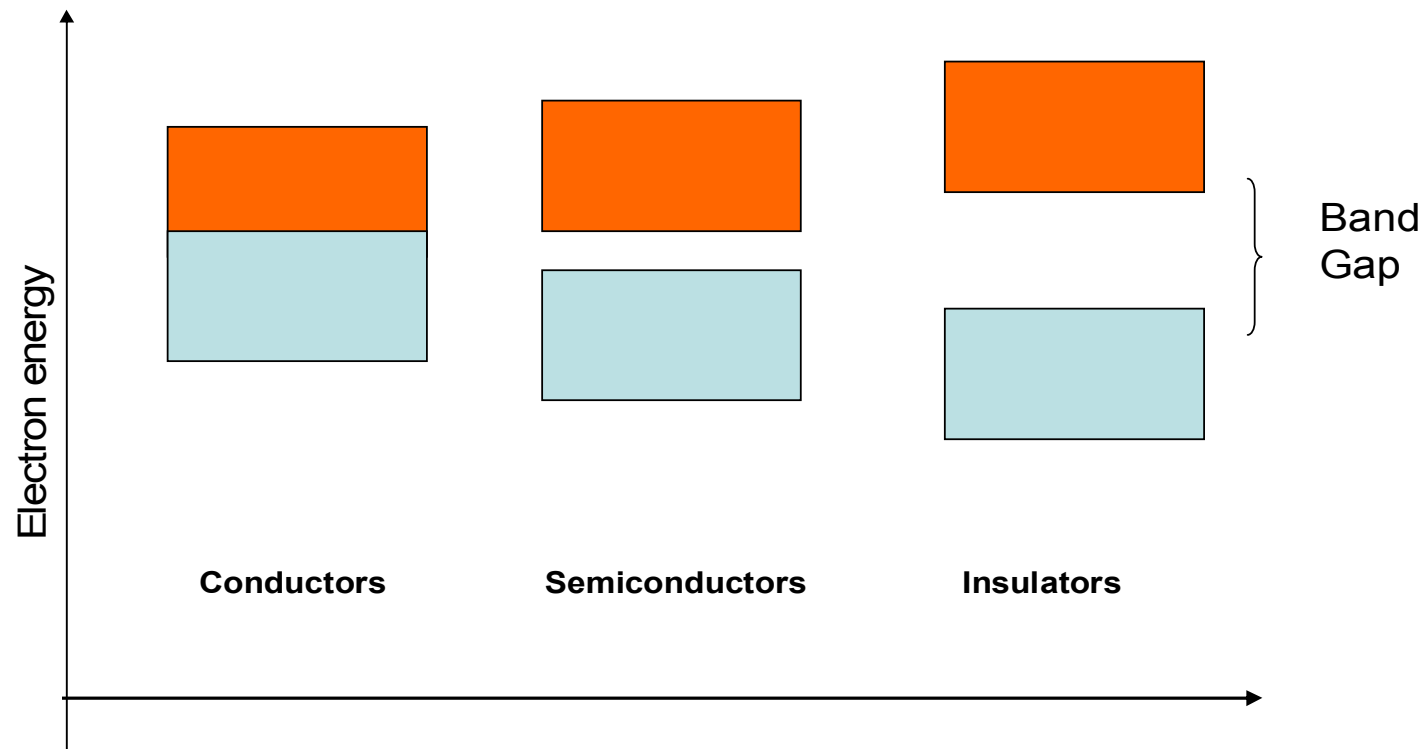
# Surface Energy & Particle Size

Side	Total Surface area	Total Edge	Surface Energy	Edge Energy
nm	sq. cm/g	cm/g	J/g	J/g
1000	$2.8 \times 10^4$	$5.5 \times 10^8$	0.56	$1.7 \times 10^{-4}$
1	$2.8 \times 10^7$	$5.5 \times 10^{14}$	560	170

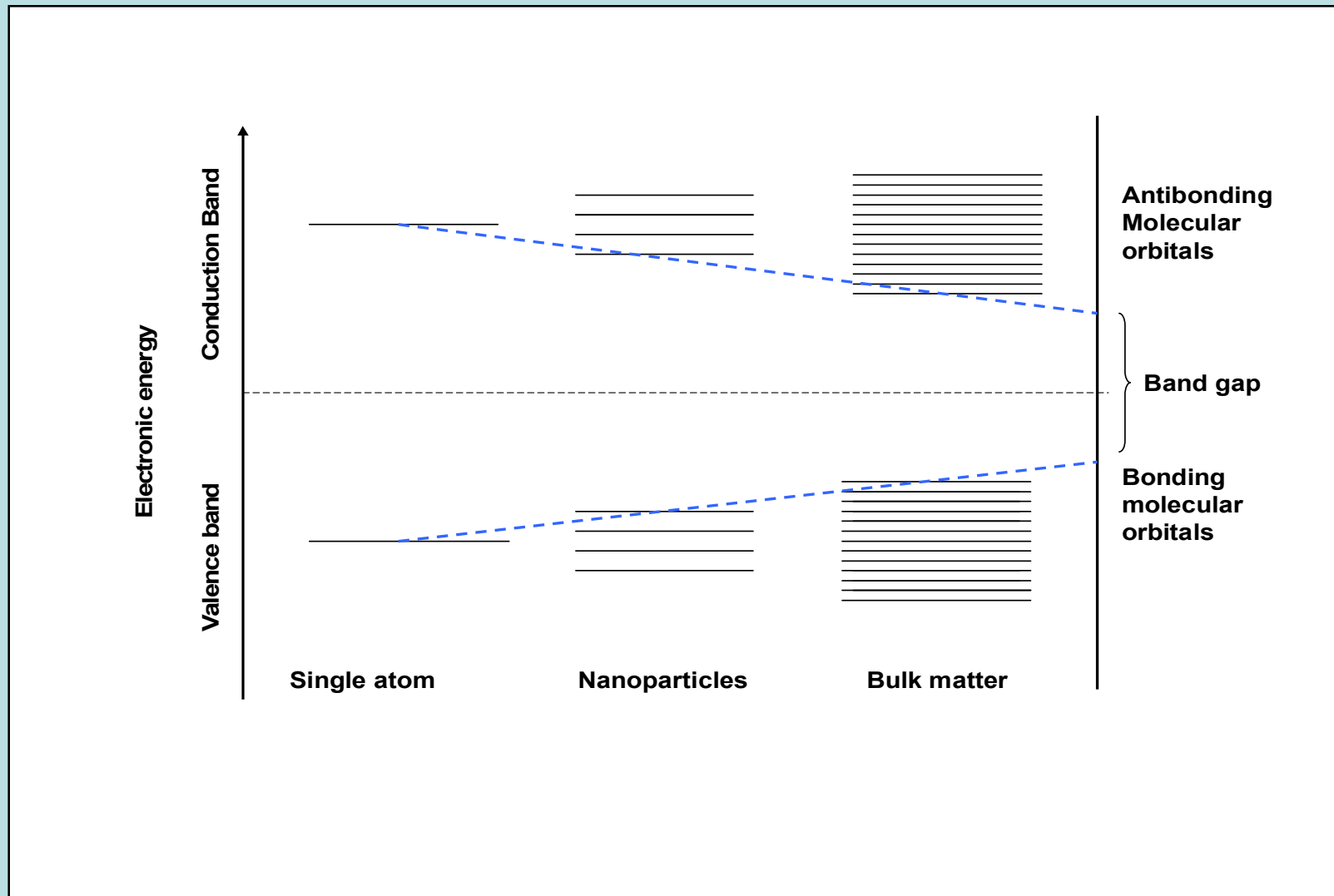
# Relaxation of High Surface Energy

1. Surface contraction, change of crystal shape and purification
2. Chemisorption
3. Self Assembly
4. Re-aggregation

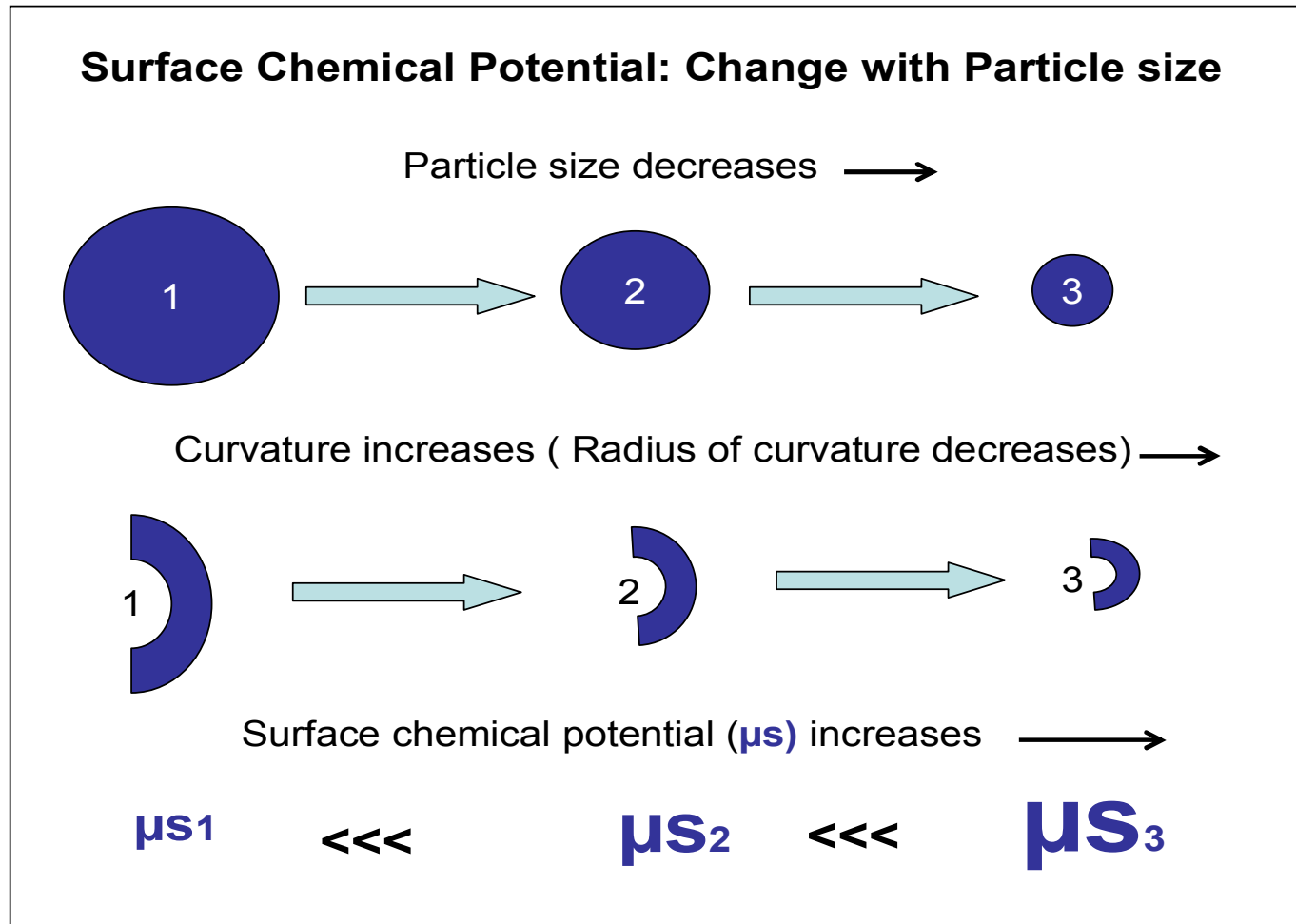
# Band Structure of Matter



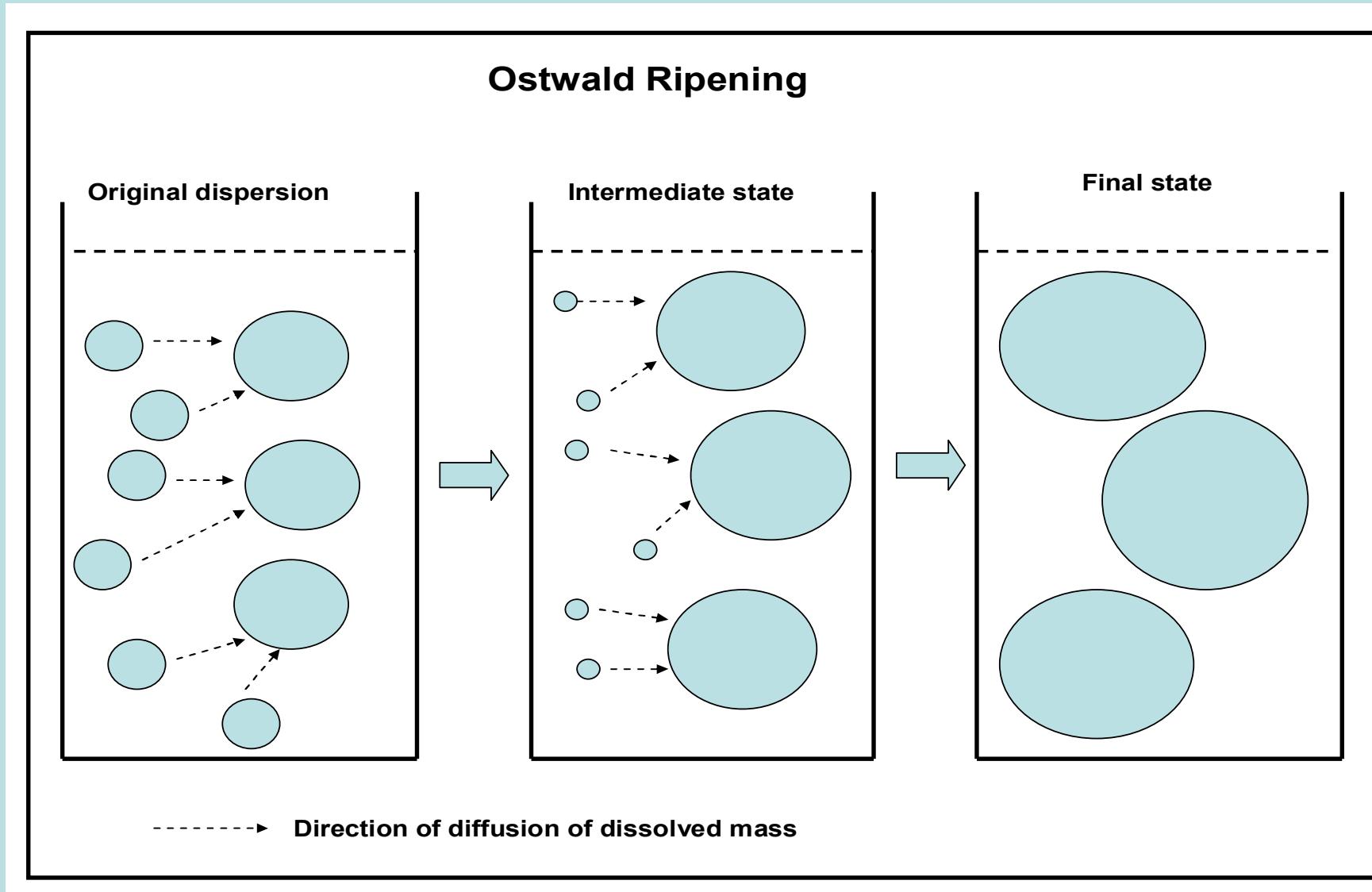
# Electronic band gap widens at nanoscale



# Surface chemical potential (reactivity) increases at nanoscale



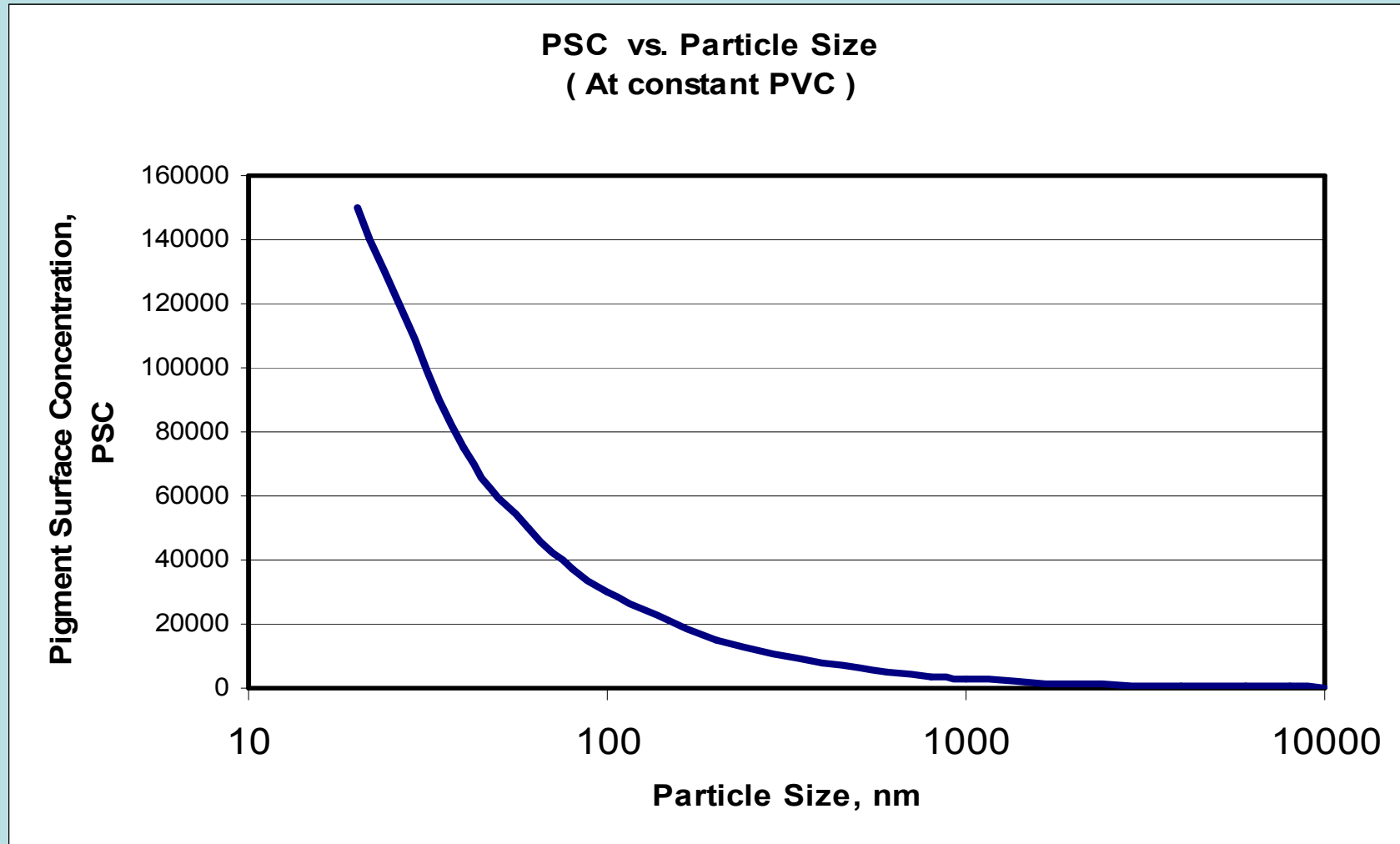
# Polydisperse particles coarsen rapidly at nanoscale



# Microdispersions vs. Nanodispersions

<u>Parameter</u>	<u>Micro-dispersion</u>	<u>Nano-dispersion</u>
<b>Particle size</b>	0.2 ~ 2.0 microns	10 ~ 70 nm
<b>Interfacial area</b>	Moderate	Very high
<b>Amount of stabilizer</b>	Moderate : 1~3 %	High 10 ~ 15 %
<b>Rheology (equal PVC)</b>	Mildly pseudoplastic	More pseudoplastic
<b>Color</b>	Original spectral	Blue shift
<b>Sedimentation</b>	Moderate. Often irreversible	Very negligible. Reversible
<b>Stability</b>	Kinetic	Thermodynamic
<b>Centrifugability</b>	Relatively easy	Difficult
<b>Examples</b>	Colourants for paints, Polymer Emulsions	Nano Inkjet printing inks, Nano polymer emulsions

High particle surface concentration requires very low dosage of nanoparticles to alter a property



# PSC & Coating Properties

- Optical: Color, Opacity, Gloss
- Photocatalytic: Light fastness
- Mechanical: Hardness, flexibility
- Barrier: Corrosion resistance, Efflorescence, Permeability
- Rheological: Flow leveling, sagging

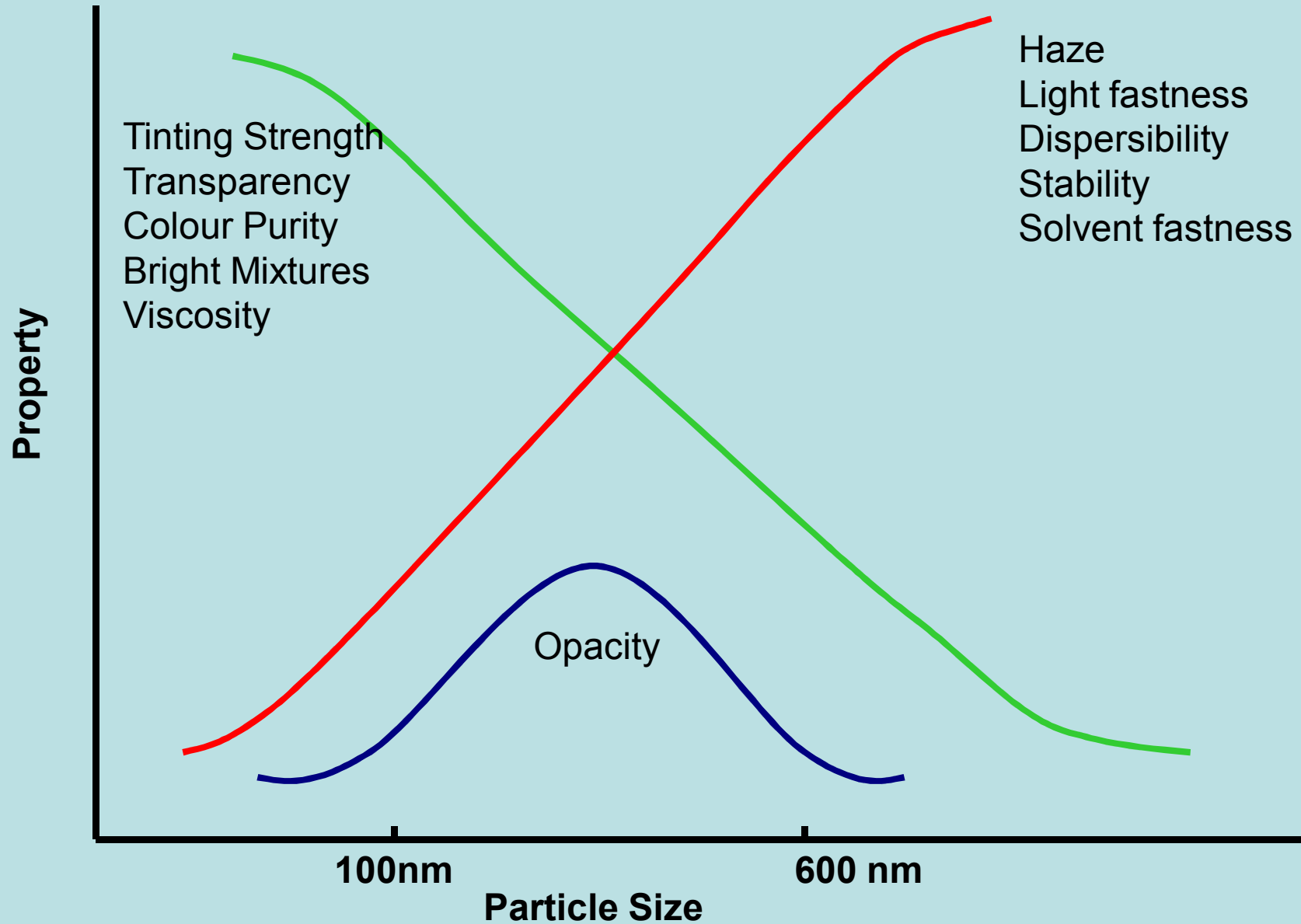
How do nanopigments differ in properties?

# Nanosize Organic & Inorganic Pigments

Conventional :100~300 nm → Nano: 30~60nm

Phenomenon	Property@ nanoscale	Effect
<b>Light Scattering</b>	Transparency	↑
	Colour purity	↑
	Haze	↓
<b>Increased Surface Chemical Potential</b>	Dispersibility	↓
	Stability	↓
	Adsorptivity	↑
	Solvent fastness	↓
<b>Increased surface area</b>	Tinting Strength	↑
	Light fastness	↓
<b>Elimination of crystal defects</b>	Photocatalysis	↑
<b>Increased band gap</b>	Hue shift	Blue shift (Angle dependent)

# Change in Properties with Particle Size



# Production Methods for Nano-Colorants

## Top Down Methods

- Nano-media mills
- Micro fluidics
- Ultrasonic dispersion
- Intercalation

## Bottom-up Methods

- Sol- gel Technique
- Reverse Micelle

# Applications of nano-colorants dispersions

- Metallic effect paints to get high flip-flop effect: Trans-iron oxides, Nano titania, HP organic pigments
- Transparent stains for wood coatings
- Candy-tone finishes for motorcycles
- Inks for ink-jet printers
- Metal decoration coatings
- Opto-electronics

## Which are the novel nano colorants?

- Coloured Metal nanoclusters
- Dye intercalated clays
- Quantum Dots
- Photonic Crystals

# Coloured Noble Metal Nanoclusters

- Colloidal gold nanoparticles are Red
- Silver nanoparticles are Yellow
- Have been used for centuries in lithography: stained glass in cathedrals
- Various other applications are being discovered: catalysis, cancer detection

# Dye Intercalated Clays

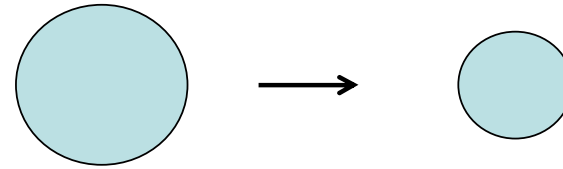
- Smectite clays can be intercalated
- Existence of a gallery or interlayer gap of 1 nm: Inserting other molecules like a dye
- Nano clay pigments: Nano clay particles intercalated with dyes act as bright colored pigments
- Good fastness properties due to shielding by inert layers (Planocolors).
- Devoid of heavy metals like lead, chromium, cadmium and mercury.

# Quantum Dots: Concept

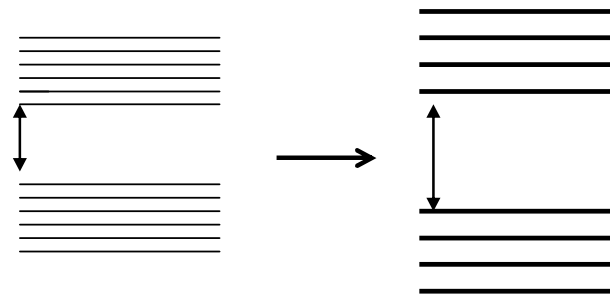
- Some of the first commercial nano-colourants
- Semiconductor nanoparticles 2~10 nm ( ~20 atoms) e.g. CdSe, ZnS, PbS
- Quantum confinement in 3D -> Fluorescence
- Absorption in UV-Vis range & emission in Vis-IR
- Emission wavelength = f ( Particle size, Distribution, Temp)
- Emission band is narrow: Bright Colours

# Change of Color with Particle Size in Quantum Dots

**Particle size decreases**



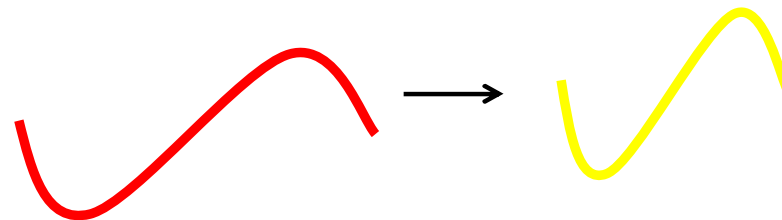
**Band gap widens & levels become discrete**



**Absorption at shorter wavelength**



**Emission at shorter wavelength : Change of color**



# Quantum Dots: Preparation & Uses

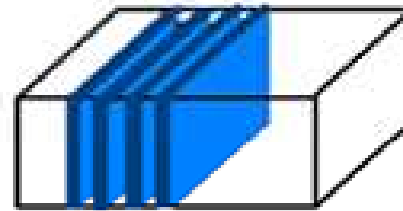
- Prepared by colloidal precipitation: Reverse Micelle
- Opto-electronics, displays, LEDs
- Photovoltaic and solar cells
- Security inks: Anti-counterfeiting
- Bright color paints: Selective highlight areas of structures, vehicles, path markings and escape-ways.
- Sighting of the distant objects like tractors in the field and highway markings.
- Night vision paints: Patrolling aid with NVG

# Photonic Band Gap (PBG) (Photonic Crystals)

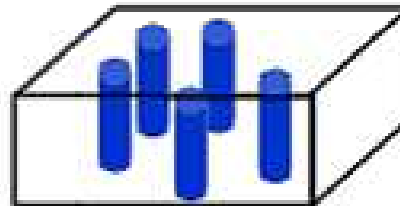
- Spatial periodicity of optically dense and rare domains like gemstone Opal
- Dimension of individual domain  $\sim$  wavelength of light
- High contrast of RI between the alternate domains (of the order of 2.5~3.0). Hence the air can be one of the domains.
- Pronounced back scatter followed by interference produces the resultant frequencies or 'colors'.
- e.g. Chromaflair pigments

# Photonic Band Gap 1D, 2D & 3D

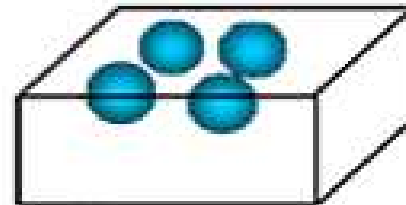
**1 Dimensional PBG**



**2 Dimensional PBG**



**3 Dimensional PBG**



# Photonic Crystals: Preparation

- Layer by layer nanofabrication
- X-ray lithography
- Laser holography
- Use of di-block copolymers
- Colloidal self assembly
- Thin film coatings

# Photonic Crystals: Applications

- 1D photonic band gap are already in use in the form of thin films like colour changing pigments: Chromaflair
- 2D photonic crystals are emerging in the area of communication through photonic crystal fibers.
- 3D photonic dielectrics : Area of intense research; applications in optical computers.

# Arthur Clarke's Third Law of Future

“Advanced technologies are  
indistinguishable from magic”

Thank you!